

Gas Breakdown and Discharge Formation in High-Power Impulse Magnetron Sputtering

Xiao Zuo¹, Rende Chen, Peiling Ke, and Aiyang Wang

Abstract—Discharge behaviors of high-power impulse magnetron sputtering with different targets have been investigated. Distinct current–voltage curves and target current waveforms are observed. Breakdown voltage and the maximum target current show a periodic drop with the increase of atomic number in subgroups and periods. The target current density is found to be mainly affected by the secondary electron emission yield. Thus, its magnitude is unable to directly evaluate the ionization degree of sputtered atoms in high-power impulse magnetron sputtering (HiPIMS) process. In this paper, the interactive influence of secondary electron emission, sputter yield, and ionization energy on the ionization degree of sputtered atoms is discussed based on the analysis of the voltage and current characteristics. As a result, targets can be categorized into three sorts according to the ionization degree: 1) low ionization degree targets, such as Ag and C less than 10%; 2) intermediate ionization degree targets like Cr and Cu with 55% and 35%; 3) Ti, Zr, and Mo targets with the second ionization processes. These results provide instructive operation ranges for the state-of-the-art HiPIMS applications.

Index Terms—Current waveform, gas breakdown, high-power impulse magnetron sputtering (HiPIMS), ionization degree, optical emission spectroscopy (OES).

I. INTRODUCTION

BENEFITS from reduced electron energy loss and high instant discharge power, such as improved plasma density, ionization degree of sputtered atoms [1], and ion

energy [2], make high-power impulse magnetron sputtering (HiPIMS) a hot topic in material engineering research studies and industrial applications. Many researchers consider it as a novel ionized physical vapor deposition technique [3] which takes advantages from dc magnetron sputtering (dcMS) and cathodic arc evaporation (CAE) in promoting substrate adhesion [4], film density [5], and surface smoothness [7], while avoiding the disadvantages from both of them like poor growth directionality, coarse columnar grain, and macroparticles. HiPIMS has already achieved great successes in microstructure modulation and property enhancement for thin films/coatings in laboratories [8]–[10]. However, its industrial application is still limited due to low deposition rate and discharge instability [11]–[13]. The loss of deposition rate attributes to the return of sputtered material ions back to target surface [14]. Meanwhile, discharge instability is also hard to avoid because HiPIMS works at abnormal glow region [16] which could easily transit into the arc region under high instant pulse voltage conditions [17]. Arcing on target surface will emit macrodroplets and degenerate thin films/coatings properties. These two features are the primary factors that need considering in HiPIMS applications.

According to specific applicable requirements, surface layers deposited by HiPIMS can be categorized into two main types: 1) surface protection coatings [18]–[20] and 2) functional thin films [7], [21], [22]. Surface protection coatings like transitional metal nitride/carbide are comprehensively deposited through dcMS or CAE methods. Their structures and properties are not quite sensitive to discharge instability unless pivotal mechanical damage caused by structural defects and property deterioration happens. Meanwhile, frequent arcing can be depressed by the advanced design of pulse unit [17]. However, the deposition rate of HiPIMS is much lower than dcMS, not even to mention CAE. Thus, although dense and refined grain nanocomposite coatings can be prepared by HiPIMS [8], [23], persuasion of coating customers turning into HiPIMS is not effective. However, low deposition rate does not matter so much to functional thin films, sometimes even becomes an advantage. Dutta *et al.* [25] reported that ultrathin Pt group metal films showed anomalous higher electric conductivity than Cu film. Film thickness can be controlled more precisely under low deposition rate conditions. Meanwhile, the high plasma density and ionization degree of sputtered atoms are beneficial to micronanoprocessing and the enhancement of film properties. For example, ions can be manipulated to fill or etch trenches and vias of semiconductor microprocessors [26], [27]. Ultrathin metal layers deposited by HiPIMS showed lower electrical resistivity than dcMS,

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75 which has promising applications in microelectronics, space,
76 and instrumentation technology [9], [28]. However, discharge
77 instability like arcing would be vital for the properties of these
78 kinds of thin films.

79 Therefore, considering the complex operation modes and
80 particle transport processes [15], [29], [30], it is necessary to
81 investigate the state-of-the-art operation ranges for various
82 HiPIMS deposition applications. Helmersson and Samuelsson
83 *et al.* [5] compared the deposition rate of eight different
84 target materials (Al, Ti, Cr, Cu, Zr, Ag, Ta, and Pt) by
85 HiPIMS with dcMS. Christie [14] analyzed the deposition
86 rate for various sputtering targets by a pathway model. The
87 ionization degree in the HiPIMS process with different targets
88 under the same conditions differs. Moreover, Yushkov and
89 Anders [6] found that gas breakdown in HiPIMS discharge
90 was a function of the time to the previous discharge pulse. The
91 discharge behaviors with various targets in HiPIMS need to be
92 clarified. Herein, we will discuss the breakdown of Ar gas and
93 discharge formation with Ag, Cu, Cr, Mo, Zr, Ti, and C targets.
94 They are widely applied in the fabrication of diamond like
95 carbon, carbon-based nanocomposites, and transitional metal
96 nitride coatings. The sputter yield and ionization energy of
97 these targets vary in large ranges. The voltage and current
98 characteristics are analyzed to find the contribution of four
99 processes on HiPIMS discharge, such as secondary electron
100 emission, gas sputter, self-sputter, and ionization of sputtered
101 materials. Especially the interactive influence of secondary
102 electron emission yield, sputter yield and ionization energy
103 on the ionization degree of sputtered materials are clarified.
104 Optical emission spectroscopy (OES) further conforms the
105 results of the analysis. Finally, a probable application scope of
106 HiPIMS deposition concerning ionization degree of sputtered
107 materials and deposition rate is suggested.

108 II. EXPERIMENTAL SETUP AND TARGET CURRENT

109 A. Experimental Setup

110 Details about the HiPIMS equipment and target current
111 measurement arrangements have been described in [16]. The
112 dimensions of the cylindrical chamber are 60 cm in diameter
113 and 60 cm in height. Background pressure for all experiments
114 is pumped to 1.5×10^{-2} mTorr to avoid the influence from
115 residual oxygen and water molecule to the utmost. Working
116 pressure is set at 3.8 mTorr by 50-sccm research grade
117 (99.999%) Ar gas. A pulse unit (HPPMS-20k, PTL) is used
118 to power the magnetron. Pulswidth and pulse frequency are
119 $200 \mu\text{s}$ and 50 Hz, respectively. The targets (99.9% purity with
120 the size of $40 \text{ cm} \times 10 \text{ cm} \times 0.7 \text{ cm}$) with various sputter
121 yields are used to investigate target current behaviors with the
122 variation of the pulse voltage. Gas sputter yield (Y_{Ar^+}) and
123 self-sputter yield (Y_{self}) under different incident energy (E_i)
124 ion bombardments are calculated by SRIM [31], which are
125 plotted in Fig. 1. OES (Acton SpectraPro SP-2500, Prince-
126 ton Instruments) is applied to characterize particle species
127 and emission intensities, which scans from 200 to 900 nm
128 with wavelength resolution of 1 nm. Although it is the
129 plasma region near the substrate holder measured by OES, its
130 results still could provide reliable references for discussion.

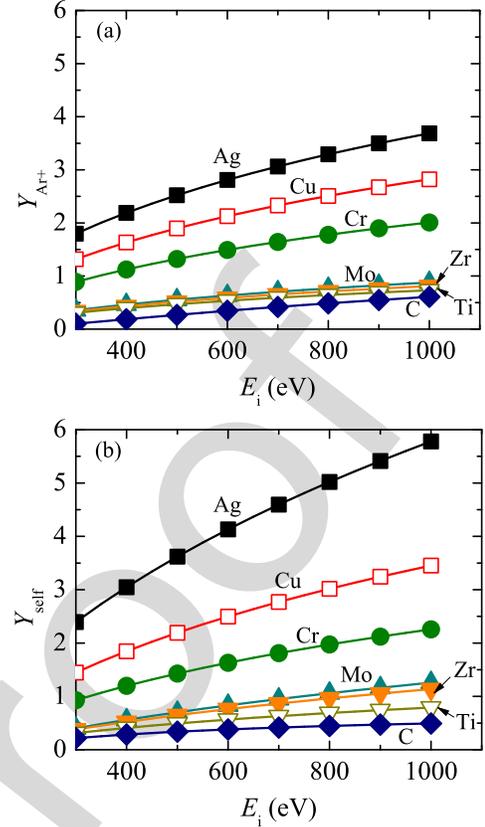


Fig. 1. (a) Sputter yield (Y_{Ar^+}) and (b) self-sputter yield (Y_{self}) of the targets under different incident ion energies (E_i). (Obtained by SRIM software.)

131 Meanwhile, it should be noted that the OES data just give
132 qualitative information on HiPIMS discharge.

133 B. About Target Current

134 Average target currents (I_{ave}) during each pulse were cal-
135 culated by the following formula:

$$136 I_{\text{ave}} = \frac{1}{T} \int_0^T I_t(t) dt \quad (1)$$

137 where $T = 20 \text{ ms}$ is the pulse period.

138 The target current generated from Ar^+ incidence (I_{Ar^+}),
139 including ion current and secondary electron current, can be
140 written in the following equation:

$$141 I_{\text{Ar}^+} = 0.5eSn_{\text{Ar}^+}(1 + \gamma_{\text{Ar}^+}) \sqrt{\frac{k_B T_e}{m_{\text{Ar}^+}}} \quad (2)$$

142 where e is the elementary charge, S is the area of racetrack,
143 n_{Ar^+} is the density of Ar^+ ions, γ_{Ar^+} is the secondary
144 electron emission yield under the bombardment of Ar^+ , k_B
145 is Boltzmann's constant, T_e is the electron temperature, and
146 m_{Ar^+} is the mass of Ar^+ . Current generated from sputtered
147 material ions ($I_{\text{M}^{z+}}$) can also be written in similar equations
148 as follows:

$$149 I_{\text{M}^{z+}} = \sum_{z=1,2} zeS\Gamma_{\text{M}^{z+}}(1 + \gamma_{\text{M}^{z+}}) \quad (3)$$

$$\Gamma_{M^+} = 0.5\alpha(Y_{Ar^+}n_{Ar^+} + Y_{self}n_{M^+} + Y_{self}'n_{M^{2+}})\sqrt{\frac{k_B T_e}{m_{M^+}}} \quad (4)$$

$$\Gamma_{M^{2+}} = 0.5\beta(Y_{Ar^+}n_{Ar^+} + Y_{self}n_{M^+} + Y_{self}'n_{M^{2+}})\sqrt{\frac{k_B T_e}{m_{M^{2+}}}} \quad (5)$$

where $\Gamma_{M^{z+}}$ is the metal ion flux to the target, α ($0 < \alpha < 1$) is the first ionization degree of the target material, β is the second ionization degree, Y_{self}' is self-sputter yield under M^{2+} ions, n_{M^+} and $n_{M^{2+}}$ are M^+ and M^{2+} density, respectively, and $m_{M^+} = m_{M^{2+}}$ is the target material ion mass. Therefore, the target current (I_t) can be obtained: $I_t = I_{Ar^+} + I_{M^{z+}}$. For simplicity, first, without considering the second ionization of sputtered material atoms, I_t can be written as the following form:

$$I_t = eS\Gamma_{Ar^+} \left(1 + \gamma_{Ar^+} + \frac{\alpha Y_{Ar^+}}{2 - \alpha Y_{self}} \sqrt{\frac{m_{Ar^+}}{m_{M^+}}} \right) \quad (6)$$

where Γ_{Ar^+} is the Ar^+ ion flux to target, which is expressed as follows:

$$\Gamma_{Ar^+} = 0.5n_{Ar^+} \sqrt{\frac{k_B T_e}{m_{M^+}}} \quad (7)$$

III. RESULTS AND DISCUSSION

A. Breakdown Voltage and Pulse Voltage Range of Different Targets in Ar HiPIMS

The current–voltage (IV) curves of HiPIMS with various targets are shown in Fig. 2. Pulse voltage is the set output voltage of the power supply, while voltage measured on the target by the oscilloscope is labeled as the target voltage. When the target voltage is high enough to ionize the Ar gas, gas breakdown and discharge occur. The breakdown voltages (U_b) of Ar gas in HiPIMS with different targets are distinct, which are plotted in Fig. 3. Breakdown voltage for graphite (C) target is the highest (721 V), while for Zr target is the lowest (356 V). The averaged target current increases with the improvement of the pulse voltage. However, too much high pulse voltage results in arc. The arcing voltages (U_{arc}) for each kind of targets are presented in Fig. 3. When the pulse voltage is higher than those values for a specific target, arcs generate. Thus, the stable work ranges of pulse voltage are obtained. Zr and Mo targets are sputtered in wider voltage ranges than others. Zr target can work in the largest voltage range, but the highest pulse voltage is achieved on the Mo target. In addition, Ti, Cu, and W targets can also work stably in large voltage ranges. However, Ag HiPIMS discharges easily transform from glow into arc.

As it is known, breakdown voltage is defined as the lowest voltage at which electric discharge occurs. It is determined by the work function (F) of the targets, which is the minimum energy needed to remove an electron from the target into the vacuum. Thus, it is expected that the variation of breakdown voltage with atomic number (Z) has a similar trend like work function. However, although the work function determines the breakdown voltage of various targets, it does not show a direct relationship with arcing voltage. In the process of glow discharge formation, energetic electrons emitted from target

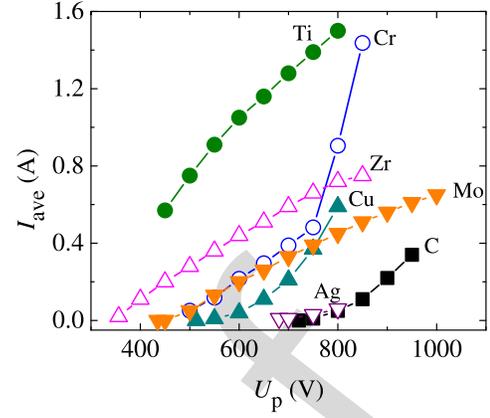


Fig. 2. Variations of averaged target current (I_{ave}) with pulse voltage (U_p) in HiPIMS discharges with different targets (measured).

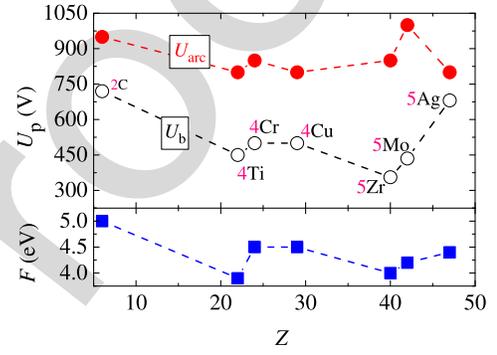


Fig. 3. Variation of breakdown voltage (U_b), arcing voltage (U_{arc}), and work function (F) with atomic number (Z).

surface ionize Ar atoms in avalanche forms. In this stage, sputtering has not happened yet. When the generated ions are attracted back to bombard target surface energetically, sputtering happens. In HiPIMS discharge, the atoms emitted from the target surface are thought to be ionized overwhelmingly as compared to dcMS [1]. This process even can dominate in HiPIMS, which is able to cause severe self-sputter [30]. Discharge state in this stage is affected by target materials greatly. Therefore, arcing voltage would be influenced by the ionization of sputtered materials.

Meanwhile, no direct relationship between target current and sputter yield is observed as shown in Fig. 2. The averaged current is the lowest on Ag target that has the highest sputter yield. When the pulse voltage is 700 V, the average current of different targets is in the order from high to low as follows: Ti, Zr, Mo, Cu, Ag, and C. In addition, with the increase in pulse voltage, IV curves present different slopes which can be classified into two groups. For example, the Cu target current increases slowly after breakdown, and then, with the increase in pulse voltage, it becomes more and more fast until arc happens. However, the increase tendency of target current with pulse voltage for Ti is on the contrary. These differences from IV curves indicate that the secondary electron emission yield, gas sputter yield, self-sputter yield, and ionization energy will affect the discharge in HiPIMS interactively.

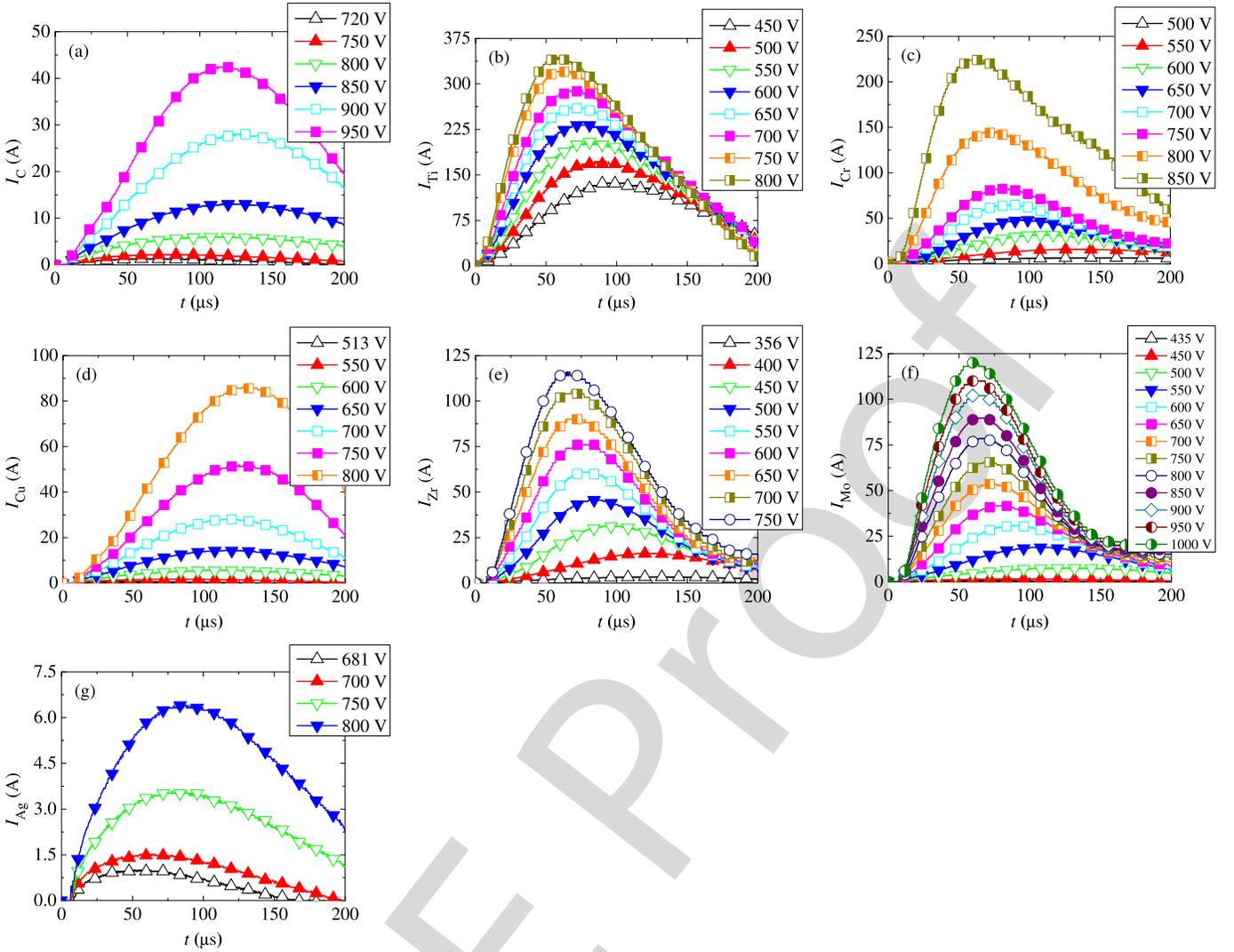


Fig. 4. Different target current waveforms in the pulse voltage ranges of stable HiPIMS discharges. Current measured on the C, Ti, Cr, Cu, Zr, Mo, and Ag targets is labeled as I_C , I_{Ti} , I_{Cr} , I_{Cu} , I_{Zr} , I_{Mo} , and I_{Ag} , respectively.

224 B. Variation of Current Waveform on Different Targets

225 The current waveforms on different targets during the
 226 200- μ s pulse-on time are demonstrated in Fig. 4. Appar-
 227 ently, all the target currents are transient. No stable current
 228 stage is observed. Although all the current waveforms present
 229 humplike shape, some details in current behaviors are distinct
 230 for different targets. First, I_{Zr} and I_{Mo} can nearly reach a
 231 stable low current stage at the end of relatively high-voltage
 232 pulse conditions (≥ 500 and 600 V, respectively). Second,
 233 the time (Δt_{max}) of target current maximum (I_p) changes with
 234 pulse voltage and differs in target elements. Δt_{max} increases
 235 with the rise of pulse voltage on Cu and Ag targets but
 236 decreases on other targets in our experiments. Third, the target
 237 current at the same pulse voltage is also different. The current
 238 waveform is interactive resultant of ionization, gas sputter,
 239 self-sputter, and gas rarefaction processes. Its evolution under
 240 various conditions has been discussed in detail by many
 241 researchers [32]–[34]. Here, we focus on the target current
 242 maximum as it is a key parameter affecting the sputtering
 243 rate. Usually, high target current is preferred in HiPIMS unless

arc generates. The highest current is obtained on Ti target
 244 in our experiments when the same pulse voltage (700 V) is
 245 applied. Details of target current waveform, including current
 246 maximum, Δt_{max} , increment rate (k_u), and decreasing rate
 247 (k_d) at 700 V for different targets, are presented in Fig. 5.
 248 The pulse voltage applied on the graphite target is selected as
 249 720 V, because 700 V is insufficient for discharge formation.
 250 For targets with transitional metal in a subgroup (III or IV)
 251 or period (third or fourth), target current maximum decreases
 252 with the increase in atomic number. Fig. 5(b) shows the
 253 change of Δt_{max} with different atomic numbers. However,
 254 Fig. 5(b) does not show a similar trend like current maxi-
 255 mum in Fig. 5(a). This could attribute to the ionization of
 256 sputtered materials. k_u and k_d are defined in the following
 257 equations:
 258

$$k_u = \frac{I_p - 0}{\Delta t_{max}} \quad (8) \quad 259$$

$$k_d = \frac{I_{end} - I_p}{200 - \Delta t_{max}} \quad (9) \quad 260$$

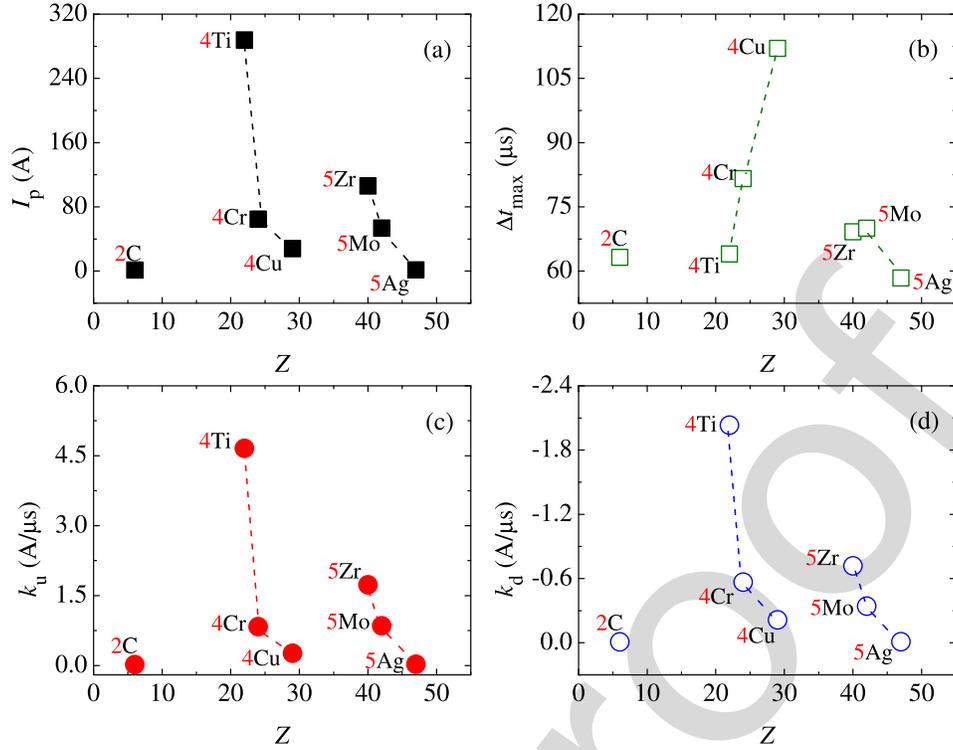


Fig. 5. Details on the current waveforms on different targets measured at with pulse voltage at 700 V, such as (a) current maximum (I_p), (b) corresponding time (Δt_{max}), (c) increase rate (k_u), and (d) decrease rate (k_d).

where I_{end} is the target current at the end of voltage pulse. As shown in (6), the rate of change in target current generally presents the ionization rate in HiPIMS discharge. It is found that for HiPIMS discharge with higher current maximum, the target current reaches maximum faster, but also decreases earlier. k_u and k_d also obey periodic feature like I_p . Generally, k_d is smaller than k_u . The change trend of them with atomic number is similar with the secondary electron emission yield except Ti.

When working pressure and pulse voltage are the same, secondary electron emission yield on different targets will lead to significant distinct in plasma density in HiPIMS. Thus, the dose of incident Ar^+ on the targets changes with different target elements. The incident Ar^+ generates new electrons and sputters target atoms out. The sputtered atoms will also be ionized in HiPIMS plasma. The target current is a sum of conductive electron current and ion current. Ion current composes of Ar^+ current and target material ions current. There would be monovalent and bivalent ions of target material according to first ionization energy ($E_{0 \rightarrow 1}$) and second ionization energy ($E_{1 \rightarrow 2}$). The electron current on the target surface is generated from Ar^+ and target material ions. The density of target material ions is influenced by Y_{Ar^+} , Y_{self} , $E_{0 \rightarrow 1}$, and $E_{1 \rightarrow 2}$. When we analyze the target current, the difference in secondary electron emission yield is considered first. Fig. 6 figures out the change of secondary electron emission yield (γ_{SEEY}) with atomic number corresponding to different target materials. The incident ion energy is assumed to be 700 eV when the pulse voltage is 700 V. As the energy per atomic mass number is less than 300 eV,

TABLE I
F, FIRST, AND SECOND IONIZATION ENERGIES FOR VARIOUS MATERIALS

Element	F (eV)	$E_{0 \rightarrow 1}$ (eV)	$E_{1 \rightarrow 2}$ (eV)
Ar	N/A	15.76	27.63
C	4.5	11.26	24.38
Ti	3.9	6.83	13.58
Cr	4.5	6.77	16.5
Cu	4.9	7.73	20.29
Zr	4.0	6.63	13.16
Mo	4.2	7.09	16.17
Ag	4.4	7.58	21.45

secondary electron emission is determined by the potential energy (E_p) of incident ions [30]. Therefore, the secondary electron yield from ion bombardment can be calculated by the following equation:

$$\gamma_{SEEY} = 0.032 * (0.78E_p - 2F). \quad (10)$$

The values of work function (F) and potential energy (E_p) for different materials are listed in Table I. As shown in Fig. 6(a), with the increase in atomic number in a subgroup or period, γ_{Ar^+} has a similar trend like the target current maximum in Fig. 5(a). It means that the electron current generated by Ar^+ incidence is dominant. However, when carefully comparing γ_{Ar^+} of the elements from the fourth period with that of the fifth period, Zr HiPIMS should have the highest target current at the same pulse voltage. However, current on Ti target is the highest; meanwhile, Cr and Cu

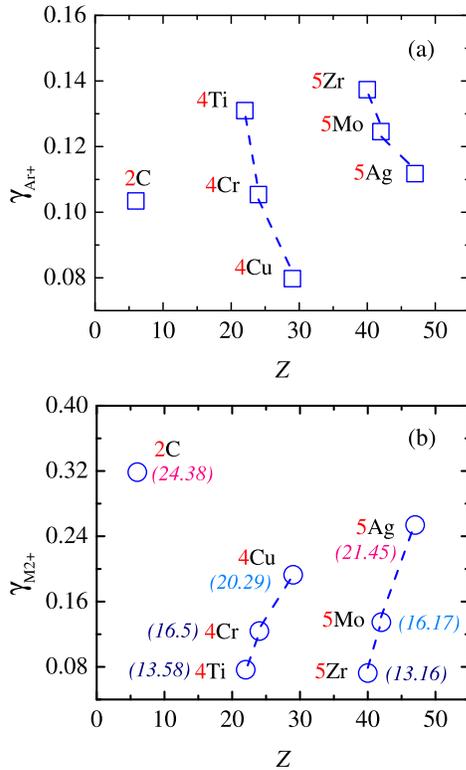


Fig. 6. Secondary electron emission yield (γ_{SEEY}) of various targets under the bombardment of (a) Ar^+ ions and (b) bivalent target material ions (M^{2+}).

306 HiPIMS also have higher target current than Mo and Ag
 307 HiPIMS, respectively. Since the monovalent ions are unable
 308 to cause secondary electron emission, the existence of bivalent
 309 target material ions is further considered. The first ionization
 310 energy of Ar is 15.76 eV, and target material atoms with
 311 the second ionization energy lower than 15.76 eV, like Ti and
 312 Zr, can be ionized into bivalent ions with high probability. In
 313 addition, there are Cr and Mo that have the second ionization
 314 energies near around 15.76 eV. These sputtered materials
 315 participated in discharge processes and are partially ionized.
 316 Therefore, different sputter yield and ionization degrees would
 317 also contribute to the variation deviation of target current
 318 maximum from γ_{Ar^+} .

319 C. Relationship Between Ionization Degree and 320 Peak Target Current

321 As the working pressure is set at 3.8 mTorr, the mean free
 322 path is larger than cathode sheath thickness. Therefore, when
 323 the pulse voltage is 700 V, it is reasonable to assume that the
 324 energy of incident ions is 700 eV as the second ionization
 325 process is neglected. Gas sputter yield and self-sputter yield
 326 of various targets with incident energy at 700 eV can be found
 327 in Fig. 1. Self-sputter yield is higher than the gas sputter yield.
 328 The difference between them is distinct according to the kind
 329 of target material. For Ag and Cu targets, the self-sputter yield
 330 is much higher than gas sputter yield when compared with
 331 others. It could be speculated that the target material with high
 332 ionization degree, self-sputter yield, and low ionization energy

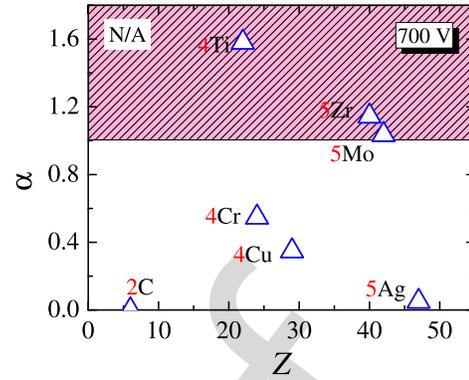


Fig. 7. Calculated ionization degree (α) of various target materials. The second ionization process exists when α is larger than 1.

333 prefers the state-of-the-art HiPIMS with high deposition rate
 334 and stability.

335 As the target current is transit, the ionization degree of
 336 sputtered materials is also expected to vary with time. The
 337 ionization degree can be calculated through the ionization
 338 region model [33]. The cross section data of excitation,
 339 ionization, deexcitation, and combination processes for different
 340 species can be found in reaction databases, such as from
 341 the OpenADAS database [35]. When the pulse voltage and
 342 working pressure are 700 V and 3.8 mTorr, the maximums of
 343 ionization degree (α) calculated from different target currents
 344 are presented in Fig. 7. According to these maximums of
 345 ionization degree, target materials can be categorized into
 346 three kinds: 1) Ag and C targets, the maximum ionization
 347 degree is less than 10%; 2) 55% and 35% for Cr and Cu;
 348 and 3) higher than 100% for Ti, Zr, and Mo. The ionization
 349 energy of C atom is the highest, thus its ionization degree
 350 is the lowest. However, the ionization energy of Ag atom
 351 is lower than that of Cu atoms, but its ionization degree
 352 is also very low. This phenomenon can be attributed to its
 353 high sputter yield, a large amount of Ag atoms participate
 354 in discharge which would reduce the electron temperature.
 355 Therefore, the target with high sputter yield will have a
 356 reduced ionization degree. Ionization degree higher than 100%
 357 is impossible, and these results are attributed to that only
 358 the first ionization processes are considered and bivalent ions
 359 are neglected. It is easy to deduce that the second ionization
 360 process exists in Ti, Zr, and Mo HiPIMS. Therefore, it is
 361 expectable to control the incident particles' energy to design
 362 dense, fine grain, and nanocomposite films for Cr, Ti, Zr, and
 363 Mo targets but obtain relatively high deposition rate for Ag
 364 and Cu targets in HiPIMS deposition process. For the high
 365 ionization energy and low sputter yield target C, other methods
 366 should be introduced to improve ionization degree [36] or
 367 deposition rate [37]. These results can also be explored to
 368 other targets according to their secondary electron emission
 369 yield, sputter yield, and ionization degree with the exception
 370 of ferromagnetic materials such as Fe, Co, and Ni.

371 D. Optical Emission Spectra of C, Cr, and 372 Ti HiPIMS Plasma

373 According to the first and second ionization of sputtered
 374 atoms, OES spectra for HiPIMS with three different targets

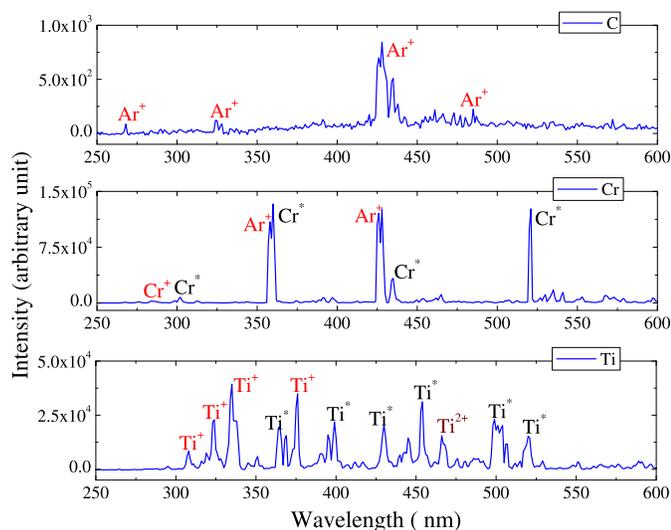


Fig. 8. OES spectra of C, Cr, and Ti HiPIMS discharge.

are detected, such as high first ionization energy element C, low second ionization energy element Ti, and high sputter yield element Cr. Correspondingly, the pulse voltage and pressure for OES measurements are 700 V and 3.8 mTorr, respectively. All the parameters of the spectroscope are set the same so that the measured OES spectra can be comparable with each other. The results are plotted in Fig. 8. C atom spectrum cannot be found. A weak spectrum peak of CrII 284 nm appears in the OES spectra of Cr HiPIMS discharge. As predicted from the ionization degree analysis, TiIII 466 nm of bivalent Ti^{2+} ions were observed in Ti HiPIMS.

IV. CONCLUSION

The discharge behaviors of HiPIMS have been investigated on Ag, Cu, Cr, Mo, Zr, Ti, and C targets. The breakdown voltage is determined by the work function of the target element, but the arcing voltage is also affected by the ionization of sputtered atoms. Stable discharge ranges of these targets are found. High Γ_{SEEY} results in low breakdown voltage and high discharge current, and vice versa. HiPIMS discharge with Ti, Mo, and Zr targets is not easy to arc at high voltage. An analytical current model was used to analyze the interactive influence of the secondary electron emission yield, sputter yield, and ionization energy on the ionization degree of sputtered atoms. The results show that the target materials with relatively low ionization energy and sputter yield, such as Cr, Ti, Zr, and Mo, tend to have higher ionization degree. Targets with high sputter yield like Ag and Cu have low ionization degree. The second ionization of sputtered atoms happens in Ti, Zr, and Mo HiPIMS. However, high sputter yield elements like Ag and Cu have relatively low ionization degree.

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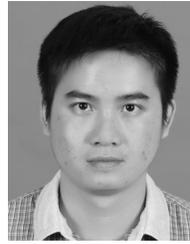
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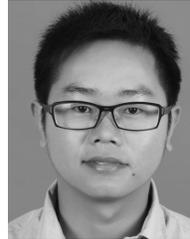
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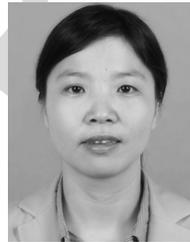
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Gas Breakdown and Discharge Formation in High-Power Impulse Magnetron Sputtering

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Abstract—Discharge behaviors of high-power impulse magnetron sputtering with different targets have been investigated. Distinct current–voltage curves and target current waveforms are observed. Breakdown voltage and the maximum target current show a periodic drop with the increase of atomic number in subgroups and periods. The target current density is found to be mainly affected by the secondary electron emission yield. Thus, its magnitude is unable to directly evaluate the ionization degree of sputtered atoms in high-power impulse magnetron sputtering (HiPIMS) process. In this paper, the interactive influence of secondary electron emission, sputter yield, and ionization energy on the ionization degree of sputtered atoms is discussed based on the analysis of the voltage and current characteristics. As a result, targets can be categorized into three sorts according to the ionization degree: 1) low ionization degree targets, such as Ag and C less than 10%; 2) intermediate ionization degree targets like Cr and Cu with 55% and 35%; 3) Ti, Zr, and Mo targets with the second ionization processes. These results provide instructive operation ranges for the state-of-the-art HiPIMS applications.

Index Terms—Current waveform, gas breakdown, high-power impulse magnetron sputtering (HiPIMS), ionization degree, optical emission spectroscopy (OES).

I. INTRODUCTION

BENEFITS from reduced electron energy loss and high instant discharge power, such as improved plasma density, ionization degree of sputtered atoms [1], and ion

energy [2], make high-power impulse magnetron sputtering (HiPIMS) a hot topic in material engineering research studies and industrial applications. Many researchers consider it as a novel ionized physical vapor deposition technique [3] which takes advantages from dc magnetron sputtering (dcMS) and cathodic arc evaporation (CAE) in promoting substrate adhesion [4], film density [5], and surface smoothness [7], while avoiding the disadvantages from both of them like poor growth directionality, coarse columnar grain, and macroparticles. HiPIMS has already achieved great successes in microstructure modulation and property enhancement for thin films/coatings in laboratories [8]–[10]. However, its industrial application is still limited due to low deposition rate and discharge instability [11]–[13]. The loss of deposition rate attributes to the return of sputtered material ions back to target surface [14]. Meanwhile, discharge instability is also hard to avoid because HiPIMS works at abnormal glow region [16] which could easily transit into the arc region under high instant pulse voltage conditions [17]. Arcing on target surface will emit macrodroplets and degenerate thin films/coatings properties. These two features are the primary factors that need considering in HiPIMS applications.

According to specific applicable requirements, surface layers deposited by HiPIMS can be categorized into two main types: 1) surface protection coatings [18]–[20] and 2) functional thin films [7], [21], [22]. Surface protection coatings like transitional metal nitride/carbide are comprehensively deposited through dcMS or CAE methods. Their structures and properties are not quite sensitive to discharge instability unless pivotal mechanical damage caused by structural defects and property deterioration happens. Meanwhile, frequent arcing can be depressed by the advanced design of pulse unit [17]. However, the deposition rate of HiPIMS is much lower than dcMS, not even to mention CAE. Thus, although dense and refined grain nanocomposite coatings can be prepared by HiPIMS [8], [23], persuasion of coating customers turning into HiPIMS is not effective. However, low deposition rate does not matter so much to functional thin films, sometimes even becomes an advantage. Dutta *et al.* [25] reported that ultrathin Pt group metal films showed anomalous higher electric conductivity than Cu film. Film thickness can be controlled more precisely under low deposition rate conditions. Meanwhile, the high plasma density and ionization degree of sputtered atoms are beneficial to micronanoprocessing and the enhancement of film properties. For example, ions can be manipulated to fill or etch trenches and vias of semiconductor microprocessors [26], [27]. Ultrathin metal layers deposited by HiPIMS showed lower electrical resistivity than dcMS,

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75 which has promising applications in microelectronics, space,
76 and instrumentation technology [9], [28]. However, discharge
77 instability like arcing would be vital for the properties of these
78 kinds of thin films.

79 Therefore, considering the complex operation modes and
80 particle transport processes [15], [29], [30], it is necessary to
81 investigate the state-of-the-art operation ranges for various
82 HiPIMS deposition applications. Helmersson and Samuelsson
83 *et al.* [5] compared the deposition rate of eight different
84 target materials (Al, Ti, Cr, Cu, Zr, Ag, Ta, and Pt) by
85 HiPIMS with dcMS. Christie [14] analyzed the deposition rate
86 for various sputtering targets by a pathway model. The
87 ionization degree in the HiPIMS process with different targets
88 under the same conditions differs. Moreover, Yushkov and
89 Anders [6] found that gas breakdown in HiPIMS discharge
90 was a function of the time to the previous discharge pulse. The
91 discharge behaviors with various targets in HiPIMS need to be
92 clarified. Herein, we will discuss the breakdown of Ar gas and
93 discharge formation with Ag, Cu, Cr, Mo, Zr, Ti, and C targets.
94 They are widely applied in the fabrication of diamond like
95 carbon, carbon-based nanocomposites, and transitional metal
96 nitride coatings. The sputter yield and ionization energy of
97 these targets vary in large ranges. The voltage and current
98 characteristics are analyzed to find the contribution of four
99 processes on HiPIMS discharge, such as secondary electron
100 emission, gas sputter, self-sputter, and ionization of sputtered
101 materials. Especially the interactive influence of secondary
102 electron emission yield, sputter yield and ionization energy
103 on the ionization degree of sputtered materials are clarified.
104 Optical emission spectroscopy (OES) further conforms the
105 results of the analysis. Finally, a probable application scope of
106 HiPIMS deposition concerning ionization degree of sputtered
107 materials and deposition rate is suggested.

108 II. EXPERIMENTAL SETUP AND TARGET CURRENT

109 A. Experimental Setup

110 Details about the HiPIMS equipment and target current
111 measurement arrangements have been described in [16]. The
112 dimensions of the cylindrical chamber are 60 cm in diameter
113 and 60 cm in height. Background pressure for all experiments
114 is pumped to 1.5×10^{-2} mTorr to avoid the influence from
115 residual oxygen and water molecule to the utmost. Working
116 pressure is set at 3.8 mTorr by 50-sccm research grade
117 (99.999%) Ar gas. A pulse unit (HPPMS-20k, PTL) is used
118 to power the magnetron. Pulswidth and pulse frequency are
119 200 μ s and 50 Hz, respectively. The targets (99.9% purity with
120 the size of 40 cm \times 10 cm \times 0.7 cm) with various sputter
121 yields are used to investigate target current behaviors with the
122 variation of the pulse voltage. Gas sputter yield (Y_{Ar^+}) and
123 self-sputter yield (Y_{self}) under different incident energy (E_i)
124 ion bombardments are calculated by SRIM [31], which are
125 plotted in Fig. 1. OES (Acton SpectraPro SP-2500, Prince-
126 ton Instruments) is applied to characterize particle species
127 and emission intensities, which scans from 200 to 900 nm
128 with wavelength resolution of 1 nm. Although it is the
129 plasma region near the substrate holder measured by OES, its
130 results still could provide reliable references for discussion.

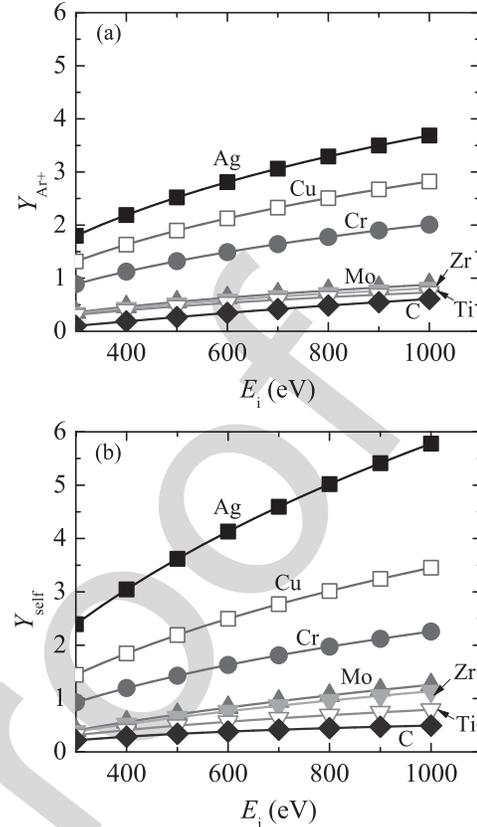


Fig. 1. (a) Sputter yield (Y_{Ar^+}) and (b) self-sputter yield (Y_{self}) of the targets under different incident ion energies (E_i). (Obtained by SRIM software.)

131 Meanwhile, it should be noted that the OES data just give
132 qualitative information on HiPIMS discharge.

133 B. About Target Current

134 Average target currents (I_{ave}) during each pulse were cal-
135 culated by the following formula:

$$136 I_{ave} = \frac{1}{T} \int_0^T I_t(t) dt \quad (1)$$

137 where $T = 20$ ms is the pulse period.

138 The target current generated from Ar^+ incidence (I_{Ar^+}),
139 including ion current and secondary electron current, can be
140 written in the following equation:

$$141 I_{Ar^+} = 0.5eSn_{Ar^+}(1 + \gamma_{Ar^+}) \sqrt{\frac{k_B T_e}{m_{Ar^+}}} \quad (2)$$

142 where e is the elementary charge, S is the area of racetrack,
143 n_{Ar^+} is the density of Ar^+ ions, γ_{Ar^+} is the secondary
144 electron emission yield under the bombardment of Ar^+ , k_B
145 is Boltzmann's constant, T_e is the electron temperature, and
146 m_{Ar^+} is the mass of Ar^+ . Current generated from sputtered
147 material ions ($I_{M^{z+}}$) can also be written in similar equations
148 as follows:

$$149 I_{M^{z+}} = \sum_{z=1,2} zeS\Gamma_{M^{z+}}(1 + \gamma_{M^{z+}}) \quad (3)$$

$$\Gamma_{M^+} = 0.5\alpha(Y_{Ar^+}n_{Ar^+} + Y_{self}n_{M^+} + Y_{self}'n_{M^{2+}})\sqrt{\frac{k_B T_e}{m_{M^+}}} \quad (4)$$

$$\Gamma_{M^{2+}} = 0.5\beta(Y_{Ar^+}n_{Ar^+} + Y_{self}n_{M^+} + Y_{self}'n_{M^{2+}})\sqrt{\frac{k_B T_e}{m_{M^{2+}}}} \quad (5)$$

where $\Gamma_{M^{z+}}$ is the metal ion flux to the target, α ($0 < \alpha < 1$) is the first ionization degree of the target material, β is the second ionization degree, Y_{self}' is self-sputter yield under M^{2+} ions, n_{M^+} and $n_{M^{2+}}$ are M^+ and M^{2+} density, respectively, and $m_{M^+} = m_{M^{2+}}$ is the target material ion mass. Therefore, the target current (I_t) can be obtained: $I_t = I_{Ar^+} + I_{M^{z+}}$. For simplicity, first, without considering the second ionization of sputtered material atoms, I_t can be written as the following form:

$$I_t = eS\Gamma_{Ar^+} \left(1 + \gamma_{Ar^+} + \frac{\alpha Y_{Ar^+}}{2 - \alpha Y_{self}} \sqrt{\frac{m_{Ar^+}}{m_{M^+}}} \right) \quad (6)$$

where Γ_{Ar^+} is the Ar^+ ion flux to target, which is expressed as follows:

$$\Gamma_{Ar^+} = 0.5n_{Ar^+} \sqrt{\frac{k_B T_e}{m_{M^+}}} \quad (7)$$

III. RESULTS AND DISCUSSION

A. Breakdown Voltage and Pulse Voltage Range of Different Targets in Ar HiPIMS

The current–voltage (IV) curves of HiPIMS with various targets are shown in Fig. 2. Pulse voltage is the set output voltage of the power supply, while voltage measured on the target by the oscilloscope is labeled as the target voltage. When the target voltage is high enough to ionize the Ar gas, gas breakdown and discharge occur. The breakdown voltages (U_b) of Ar gas in HiPIMS with different targets are distinct, which are plotted in Fig. 3. Breakdown voltage for graphite (C) target is the highest (721 V), while for Zr target is the lowest (356 V). The averaged target current increases with the improvement of the pulse voltage. However, too much high pulse voltage results in arc. The arcing voltages (U_{arc}) for each kind of targets are presented in Fig. 3. When the pulse voltage is higher than those values for a specific target, arcs generate. Thus, the stable work ranges of pulse voltage are obtained. Zr and Mo targets are sputtered in wider voltage ranges than others. Zr target can work in the largest voltage range, but the highest pulse voltage is achieved on the Mo target. In addition, Ti, Cu, and W targets can also work stably in large voltage ranges. However, Ag HiPIMS discharges easily transform from glow into arc.

As it is known, breakdown voltage is defined as the lowest voltage at which electric discharge occurs. It is determined by the work function (F) of the targets, which is the minimum energy needed to remove an electron from the target into the vacuum. Thus, it is expected that the variation of breakdown voltage with atomic number (Z) has a similar trend like work function. However, although the work function determines the breakdown voltage of various targets, it does not show a direct relationship with arcing voltage. In the process of glow discharge formation, energetic electrons emitted from target

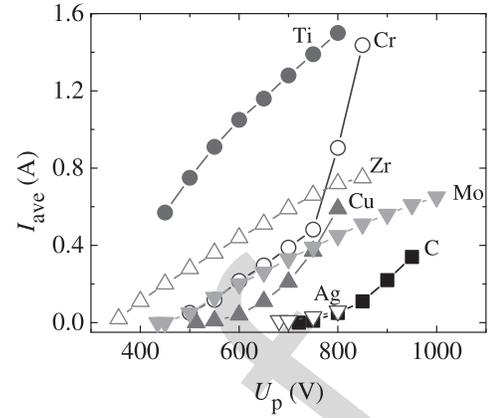


Fig. 2. Variations of averaged target current (I_{ave}) with pulse voltage (U_p) in HiPIMS discharges with different targets (measured).

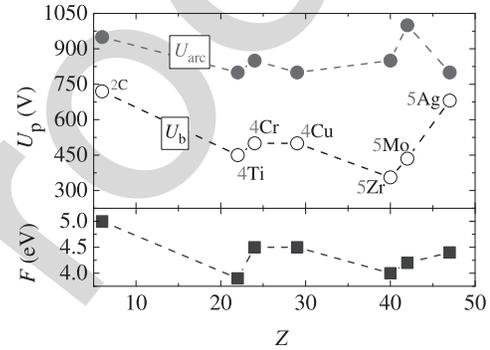


Fig. 3. Variation of breakdown voltage (U_b), arcing voltage (U_{arc}), and work function (F) with atomic number (Z).

surface ionize Ar atoms in avalanche forms. In this stage, sputtering has not happened yet. When the generated ions are attracted back to bombard target surface energetically, sputtering happens. In HiPIMS discharge, the atoms emitted from the target surface are thought to be ionized overwhelmingly as compared to dcMS [1]. This process even can dominate in HiPIMS, which is able to cause severe self-sputter [30]. Discharge state in this stage is affected by target materials greatly. Therefore, arcing voltage would be influenced by the ionization of sputtered materials.

Meanwhile, no direct relationship between target current and sputter yield is observed as shown in Fig. 2. The averaged current is the lowest on Ag target that has the highest sputter yield. When the pulse voltage is 700 V, the average current of different targets is in the order from high to low as follows: Ti, Zr, Mo, Cu, Ag, and C. In addition, with the increase in pulse voltage, IV curves present different slopes which can be classified into two groups. For example, the Cu target current increases slowly after breakdown, and then, with the increase in pulse voltage, it becomes more and more fast until arc happens. However, the increase tendency of target current with pulse voltage for Ti is on the contrary. These differences from IV curves indicate that the secondary electron emission yield, gas sputter yield, self-sputter yield, and ionization energy will affect the discharge in HiPIMS interactively.

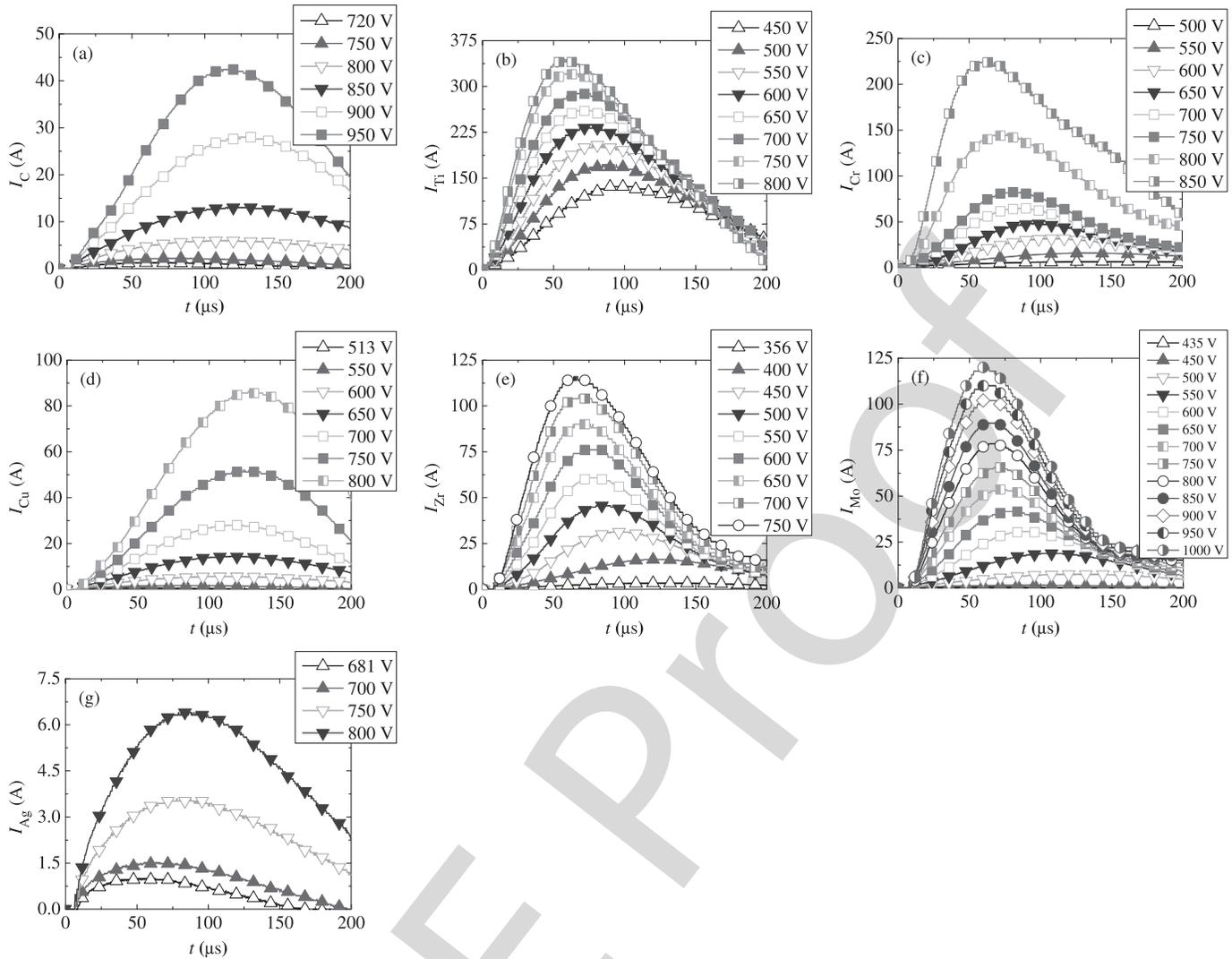


Fig. 4. Different target current waveforms in the pulse voltage ranges of stable HiPIMS discharges. Current measured on the C, Ti, Cr, Cu, Zr, Mo, and Ag targets is labeled as I_C , I_{Ti} , I_{Cr} , I_{Cu} , I_{Zr} , I_{Mo} , and I_{Ag} , respectively.

224 B. Variation of Current Waveform on Different Targets

225 The current waveforms on different targets during the
 226 200- μ s pulse-on time are demonstrated in Fig. 4. Appar-
 227 ently, all the target currents are transient. No stable current
 228 stage is observed. Although all the current waveforms present
 229 humplike shape, some details in current behaviors are distinct
 230 for different targets. First, I_{Zr} and I_{Mo} can nearly reach a
 231 stable low current stage at the end of relatively high-voltage
 232 pulse conditions (≥ 500 and 600 V, respectively). Second,
 233 the time (Δt_{max}) of target current maximum (I_p) changes with
 234 pulse voltage and differs in target elements. Δt_{max} increases
 235 with the rise of pulse voltage on Cu and Ag targets but
 236 decreases on other targets in our experiments. Third, the target
 237 current at the same pulse voltage is also different. The current
 238 waveform is interactive resultant of ionization, gas sputter,
 239 self-sputter, and gas rarefaction processes. Its evolution under
 240 various conditions has been discussed in detail by many
 241 researchers [32]–[34]. Here, we focus on the target current
 242 maximum as it is a key parameter affecting the sputtering
 243 rate. Usually, high target current is preferred in HiPIMS unless

arc generates. The highest current is obtained on Ti target
 in our experiments when the same pulse voltage (700 V) is
 applied. Details of target current waveform, including current
 maximum, Δt_{max} , increment rate (k_u), and decreasing rate
 (k_d) at 700 V for different targets, are presented in Fig. 5.
 The pulse voltage applied on the graphite target is selected as
 720 V, because 700 V is insufficient for discharge formation.
 For targets with transitional metal in a subgroup (III or IV)
 or period (third or fourth), target current maximum decreases
 with the increase in atomic number. Fig. 5(b) shows the
 change of Δt_{max} with different atomic numbers. However,
 Fig. 5(b) does not show a similar trend like current maxi-
 mum in Fig. 5(a). This could attribute to the ionization of
 sputtered materials. k_u and k_d are defined in the following
 equations:

$$k_u = \frac{I_p - 0}{\Delta t_{max}} \quad (8) \quad 259$$

$$k_d = \frac{I_{end} - I_p}{200 - \Delta t_{max}} \quad (9) \quad 260$$

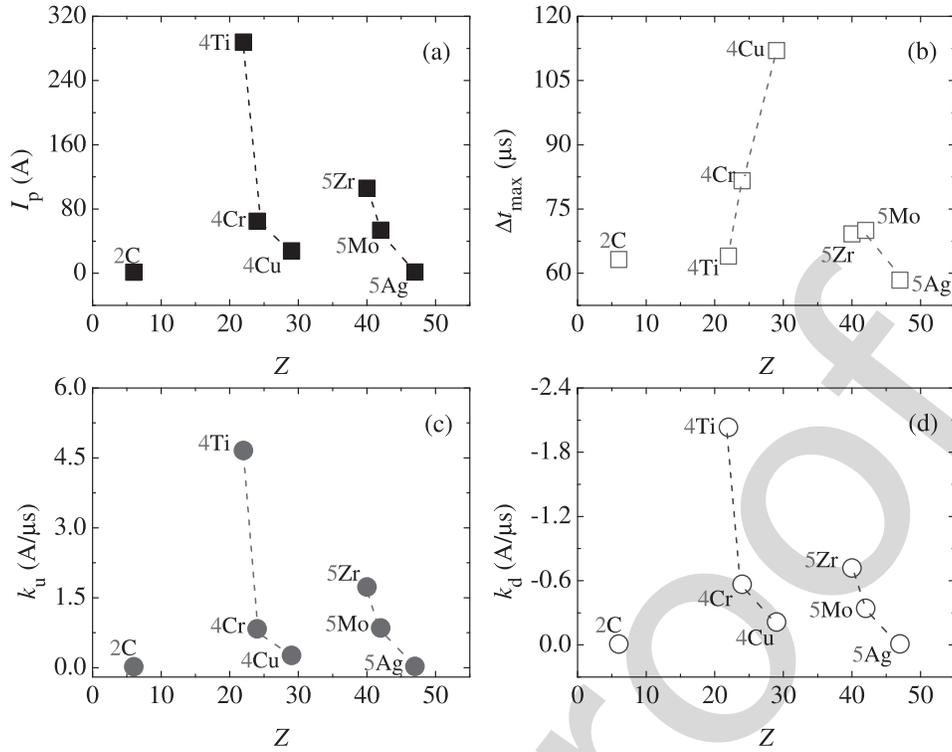


Fig. 5. Details on the current waveforms on different targets measured at with pulse voltage at 700 V, such as (a) current maximum (I_p), (b) corresponding time (Δt_{max}), (c) increase rate (k_u), and (d) decrease rate (k_d).

where I_{end} is the target current at the end of voltage pulse. As shown in (6), the rate of change in target current generally presents the ionization rate in HiPIMS discharge. It is found that for HiPIMS discharge with higher current maximum, the target current reaches maximum faster, but also decreases earlier. k_u and k_d also obey periodic feature like I_p . Generally, k_d is smaller than k_u . The change trend of them with atomic number is similar with the secondary electron emission yield except Ti.

When working pressure and pulse voltage are the same, secondary electron emission yield on different targets will lead to significant distinct in plasma density in HiPIMS. Thus, the dose of incident Ar^+ on the targets changes with different target elements. The incident Ar^+ generates new electrons and sputters target atoms out. The sputtered atoms will also be ionized in HiPIMS plasma. The target current is a sum of conductive electron current and ion current. Ion current composes of Ar^+ current and target material ions current. There would be monovalent and bivalent ions of target material according to first ionization energy ($E_{0 \rightarrow 1}$) and second ionization energy ($E_{1 \rightarrow 2}$). The electron current on the target surface is generated from Ar^+ and target material ions. The density of target material ions is influenced by Y_{Ar^+} , Y_{self} , $E_{0 \rightarrow 1}$, and $E_{1 \rightarrow 2}$. When we analyze the target current, the difference in secondary electron emission yield is considered first. Fig. 6 figures out the change of secondary electron emission yield (γ_{SEEY}) with atomic number corresponding to different target materials. The incident ion energy is assumed to be 700 eV when the pulse voltage is 700 V. As the energy per atomic mass number is less than 300 eV,

TABLE I
F, FIRST, AND SECOND IONIZATION ENERGIES FOR VARIOUS MATERIALS

Element	F (eV)	$E_{0 \rightarrow 1}$ (eV)	$E_{1 \rightarrow 2}$ (eV)
Ar	N/A	15.76	27.63
C	4.5	11.26	24.38
Ti	3.9	6.83	13.58
Cr	4.5	6.77	16.5
Cu	4.9	7.73	20.29
Zr	4.0	6.63	13.16
Mo	4.2	7.09	16.17
Ag	4.4	7.58	21.45

secondary electron emission is determined by the potential energy (E_p) of incident ions [30]. Therefore, the secondary electron yield from ion bombardment can be calculated by the following equation:

$$\gamma_{SEEY} = 0.032 * (0.78E_p - 2F). \quad (10)$$

The values of work function (F) and potential energy (E_p) for different materials are listed in Table I. As shown in Fig. 6(a), with the increase in atomic number in a subgroup or period, γ_{Ar^+} has a similar trend like the target current maximum in Fig. 5(a). It means that the electron current generated by Ar^+ incidence is dominant. However, when carefully comparing γ_{Ar^+} of the elements from the fourth period with that of the fifth period, Zr HiPIMS should have the highest target current at the same pulse voltage. However, current on Ti target is the highest; meanwhile, Cr and Cu

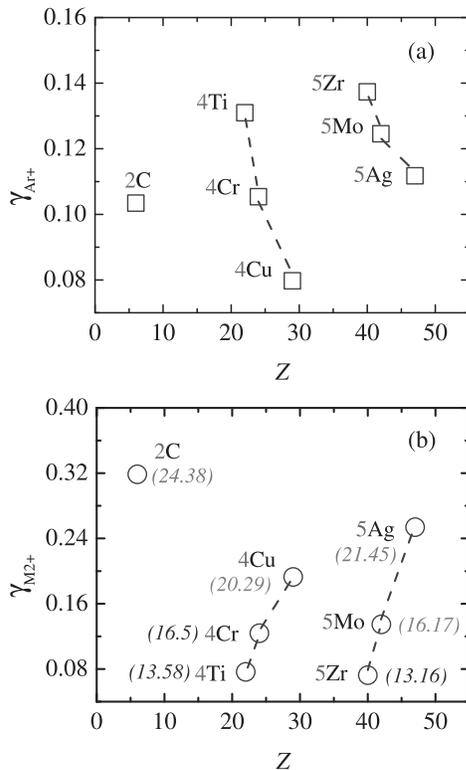


Fig. 6. Secondary electron emission yield (γ_{SEEEY}) of various targets under the bombardment of (a) Ar^+ ions and (b) bivalent target material ions (M^{2+}).

306 HiPIMS also have higher target current than Mo and Ag
 307 HiPIMS, respectively. Since the monovalent ions are unable
 308 to cause secondary electron emission, the existence of bivalent
 309 target material ions is further considered. The first ionization
 310 energy of Ar is 15.76 eV, and target material atoms with
 311 the second ionization energy lower than 15.76 eV, like Ti and
 312 Zr, can be ionized into bivalent ions with high probability. In
 313 addition, there are Cr and Mo that have the second ionization
 314 energies near around 15.76 eV. These sputtered materials
 315 participated in discharge processes and are partially ionized.
 316 Therefore, different sputter yield and ionization degrees would
 317 also contribute to the variation deviation of target current
 318 maximum from γ_{Ar^+} .

319 C. Relationship Between Ionization Degree and 320 Peak Target Current

321 As the working pressure is set at 3.8 mTorr, the mean free
 322 path is larger than cathode sheath thickness. Therefore, when
 323 the pulse voltage is 700 V, it is reasonable to assume that the
 324 energy of incident ions is 700 eV as the second ionization
 325 process is neglected. Gas sputter yield and self-sputter yield
 326 of various targets with incident energy at 700 eV can be found
 327 in Fig. 1. Self-sputter yield is higher than the gas sputter yield.
 328 The difference between them is distinct according to the kind
 329 of target material. For Ag and Cu targets, the self-sputter yield
 330 is much higher than gas sputter yield when compared with
 331 others. It could be speculated that the target material with high
 332 ionization degree, self-sputter yield, and low ionization energy

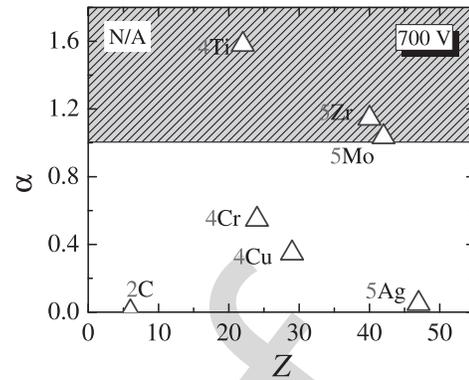


Fig. 7. Calculated ionization degree (α) of various target materials. The second ionization process exists when α is larger than 1.

333 prefers the state-of-the-art HiPIMS with high deposition rate
 334 and stability.

335 As the target current is transit, the ionization degree of
 336 sputtered materials is also expected to vary with time. The
 337 ionization degree can be calculated through the ionization
 338 region model [33]. The cross section data of excitation,
 339 ionization, deexcitation, and combination processes for different
 340 species can be found in reaction databases, such as from
 341 the OpenADAS database [35]. When the pulse voltage and
 342 working pressure are 700 V and 3.8 mTorr, the maximums of
 343 ionization degree (α) calculated from different target currents
 344 are presented in Fig. 7. According to these maximums of
 345 ionization degree, target materials can be categorized into
 346 three kinds: 1) Ag and C targets, the maximum ionization
 347 degree is less than 10%; 2) 55% and 35% for Cr and Cu;
 348 and 3) higher than 100% for Ti, Zr, and Mo. The ionization
 349 energy of C atom is the highest, thus its ionization degree
 350 is the lowest. However, the ionization energy of Ag atom
 351 is lower than that of Cu atoms, but its ionization degree
 352 is also very low. This phenomenon can be attributed to its
 353 high sputter yield, a large amount of Ag atoms participate
 354 in discharge which would reduce the electron temperature.
 355 Therefore, the target with high sputter yield will have a
 356 reduced ionization degree. Ionization degree higher than 100%
 357 is impossible, and these results are attributed to that only
 358 the first ionization processes are considered and bivalent ions
 359 are neglected. It is easy to deduce that the second ionization
 360 process exists in Ti, Zr, and Mo HiPIMS. Therefore, it is
 361 expectable to control the incident particles' energy to design
 362 dense, fine grain, and nanocomposite films for Cr, Ti, Zr, and
 363 Mo targets but obtain relatively high deposition rate for Ag
 364 and Cu targets in HiPIMS deposition process. For the high
 365 ionization energy and low sputter yield target C, other methods
 366 should be introduced to improve ionization degree [36] or
 367 deposition rate [37]. These results can also be explored to
 368 other targets according to their secondary electron emission
 369 yield, sputter yield, and ionization degree with the exception
 370 of ferromagnetic materials such as Fe, Co, and Ni.

371 D. Optical Emission Spectra of C, Cr, and 372 Ti HiPIMS Plasma

373 According to the first and second ionization of sputtered
 374 atoms, OES spectra for HiPIMS with three different targets

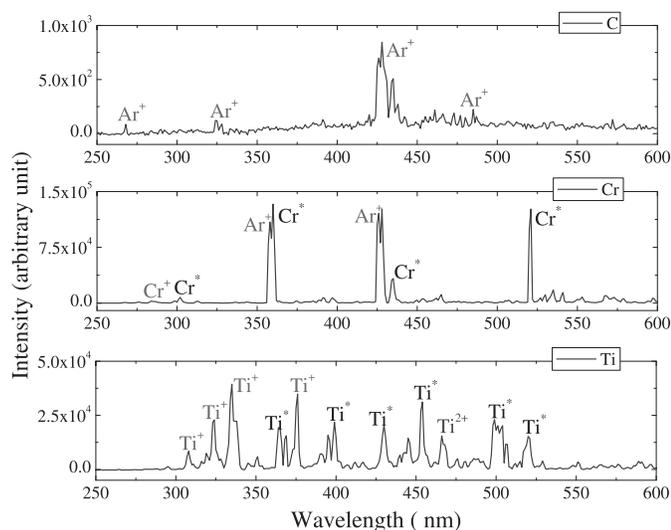


Fig. 8. OES spectra of C, Cr, and Ti HiPIMS discharge.

are detected, such as high first ionization energy element C, low second ionization energy element Ti, and high sputter yield element Cr. Correspondingly, the pulse voltage and pressure for OES measurements are 700 V and 3.8 mTorr, respectively. All the parameters of the spectroscope are set the same so that the measured OES spectra can be comparable with each other. The results are plotted in Fig. 8. C atom spectrum cannot be found. A weak spectrum peak of CrII 284 nm appears in the OES spectra of Cr HiPIMS discharge. As predicted from the ionization degree analysis, TiIII 466 nm of bivalent Ti^{2+} ions were observed in Ti HiPIMS.

IV. CONCLUSION

The discharge behaviors of HiPIMS have been investigated on Ag, Cu, Cr, Mo, Zr, Ti, and C targets. The breakdown voltage is determined by the work function of the target element, but the arcing voltage is also affected by the ionization of sputtered atoms. Stable discharge ranges of these targets are found. High Γ_{SEEY} results in low breakdown voltage and high discharge current, and vice versa. HiPIMS discharge with Ti, Mo, and Zr targets is not easy to arc at high voltage. An analytical current model was used to analyze the interactive influence of the secondary electron emission yield, sputter yield, and ionization energy on the ionization degree of sputtered atoms. The results show that the target materials with relatively low ionization energy and sputter yield, such as Cr, Ti, Zr, and Mo, tend to have higher ionization degree. Targets with high sputter yield like Ag and Cu have low ionization degree. The second ionization of sputtered atoms happens in Ti, Zr, and Mo HiPIMS. However, high sputter yield elements like Ag and Cu have relatively low ionization degree.

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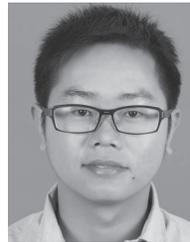
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